## THE FULL STRUCTURES OF THREE CHROMOMYCINS,

 $A_2$ ,  $A_3$  and  $A_4$ 

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The chromomycins are a group of cancerostatic antibiotics produced by Streptomyces griseus<sup>1)</sup> and are undoubtedly closely related to olivomycin<sup>2)</sup> and several other antibiotics<sup>3)</sup> isolated by various groups. The principal constituent, chromomycin  $A_3(\underline{1})$  is commercially available<sup>4)</sup>, while chromomycin  $A_2(\underline{2})$  and chromomycin  $A_4(\underline{3})$  are the major byproducts isolated during the manufacture of  $A_3$ . The structures of these three constituents are reported in this communication.

The three chromomycins<sup>5)</sup> can be isolated by silica gel chromatography of the chromomycin A mixture using ethyl acetate containing 1% oxalic acid as the solvent, the ratio of  $A_2$ ,  $A_3$  and  $A_4$  being roughly 1:8:1.

1] Chromomycin A<sub>3</sub> (1)

A planar structure  $\underline{4}$  had previously been derived <sup>6)</sup> for chromomycin  $A_3$  on grounds of a "tetrasaccharide" that was isolated from the alkaline hydrolysate of its "methyl ether," m. p. 185°. However, subsequent

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6) monodeacetylchromomycin A<sub>3</sub> R: deacetylchromose B

investigations disclosed that the methyl ether, although crystalline, was a mixture of several constituents, and therefore the linkage and sequence of the four sugars, chromoses  $A^{(8)(9)}$ ,  $B^{(10)}$ ,  $C^{(10)}$  and  $D^{(10)(11)}$  were reinvestigated since the exact nature of the "tetrasaccharide" became dubious. This reinvestigation, together with elucidation of the full stereochemistry of the chromophore, chromomycinone<sup>(12)</sup> (5) has now led to expression 1 (Chart 1).



Table 1. Chemical shifts of acetoxyl and C<sub>7</sub>-Me peaks of chromomycinone (5) derivatives (ppm from internal TMS).

Position	C-9	C-8	C-6	C-2	C-3	Me*	C-4'
Average	2. 5	2.4	2.35	2. 3	2, 2	2. 1	2. 0
Hexaacetate	2. 47	2. 41	2, 35	2, 28	2. 23	2.10	2. 01
8, 3', 4'-tri- acetate		2. 41			2. 21	2, 05	1, 97

\* The aromatic methyl peaks are easily characterized because of their low height arising from coupling to the 5-H.

Table 2. Chemical shifts of carbinyl protons (ppm)

Position	. 2	3 1	4
Hexaacetate	5, 57	5. 28	5. 42
8, 3', 4'-triacetate	4. 28	5.33	5, 45

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The points of attachment of sugar residues to chromomycinone were deduced in the following manner. An extensive comparison of the NMR acetoxyl peaks of thirteen chromomycinone derivatives acetylated at various positions revealed that the six acetoxyls, attached to C-9, C-8, C-6, C-2, C-3 and C-4, respectively, could be easily differentiated because each of these acetoxyl groups absorbed within a very narrow range characteristic of its position in the molecule (Table 1). Controlled hydrolysis of A3-peracetate with 50% formic acid yielded a chromomycinone triacetate. The chemical shifts of the acetoxyl peaks of this triacetate (Table 1) unambiguously showed that positions C-9, C-6 and C-2 are unacetylated; the same conclusion is derived by comparing the chemical shifts of the carbinyl protons (Table 2), from which it is obvious that C2-OH is unacetylated. These three positions indicate possible points of attachment, but C-9 can be excluded since the NMR spectrum of  $A_2$  itself has two clear peaks at 9.79 and 15.75 ppm, which can be assigned to the C-8 and C-9 hydroxyls<sup>6)</sup>. Thus the C-2 and C-6 are the only two possible positions for linkage of the sugars. In addition, it has already been shown that chromose D is attached to  $C-6^{6}$ 

Careful hydrolysis of  $A_3$  with aqueous methanolic potassium carbonate yielded monoceacetylchromomycin  $A_3$  (6) (Chart 1), which upon further hydrolysis with 50% acetic acid yielded CHR-D-C-A (7) (Chart 1)<sup>7)</sup>. Since chromose B is not contained in this partial hydrolysis product it must constitute a terminal residue.

Methanolysis of  $A_3$ -pentatosylate (all hydroxyls excepting  $C_9$ -OH are tosylated; ev.dence from NMR) afforded the methyl glycosides of

chromose B, chromose D, deacetyl chromose D, 3-O-tosylchromose  $A^{13}$ and 4-O-tosylchromose  $C^{13}$ ; production of 3-O-tosylchromose A indicates that chromose A must be at a terminal position, while production of 4-O-tosylchromose C indicates that of the two hydroxyl groups present in chromose C, at least the  $C_4$ -hydroxyl must have been free in the original antibiotic.

The above evidence reduces the number of possible structures for  ${\rm A}^{}_3$  to the following four:

I.	A-D-C-B
II.	B-D-CHR -C-A
111.	B-C-D- <sup>2</sup> CHR - A
IV.	A-C-D- <sup>2</sup> CHR <sub>6</sub> -B

The numerals, 2 and 6, indicate positions to which sugars are linked.

However, it was found that methanolysis of CHR-D-C-B pentatosylate (all hydroxyls excepting  $C_9$ -OH are tosylated) afforded the methyl glycosides of 4-O-tosylchromose  $C^{13}$ , 4-O-tosyldeacetylchromose  $D^{13}$ ) and chromose B. This result can only be accounted for by sequence I. Finally, because a hemiketal is easily formed between the 2-OH and 2'-CO in chromomycinone derivatives <sup>12)</sup>, a further choice had to be made between the two possibilities Ia and Ib in sequence I.



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However, Ib could be eliminated because the IR spectrum (KBr disk) of CHR-D-C-A (7) clearly showed three absorptions in the carbonyl region at 1738 (OAc of chromose D), 1728 (side-chain CO) and 1630 cm<sup>-1</sup> (annular CO). The total structure of chromomycin  $A_3$  is thus represented by <u>1</u> (corresponding to a molecular formula of  $C_{51}H_{72}O_{23}$ )<sup>14)</sup>

## 2] <u>Chromomycin A</u><sub>2</sub> (2)

Chromomycin  $A_2$  was obtained as a yellow powder with  $[a]_D^{23} - 61^\circ$ (c 1.0 in ethanol),  $\lambda_{max}^{EtOH} m\mu (\log \epsilon) 229 (4.37)$ , 279 (4.67), 317 (3.86), 331 (3.75, 412 (3.89), and  $v^{CHCl_3}$  3400, 1730, 1715 (shoulder), 1630, 1205, 1065 cm<sup>-1</sup>. The spectroscopic properties are very similar to chromomycin  $A_3$  excepting that slight differences are detected in the NMR spectra. Namely, in the  $A_2$  spectrum, the intensity of methyl peaks around 1.3 ppm is greater, and there is only one acetoxyl peak around 2 ppm (in contrast,  $A_3$  has two ac etoxyl peaks).

As mentioned above, careful hydrolysis of  $A_3$  afforded monodeacetylchromomycin  $A_3$  (6) (Chart 1). This same product (analyses, IR, UV, rotation) was obtained when  $A_2$  was submitted to identical hydrolytic conditions. On the other hand, methanolysis of  $A_2$  pentatosylate gave the methyl glycosides of deacetylchromose D, 3-O-tosylchromose A, 4-O-tosylchromose C, and chromose B'. Further hydrolysis of methyl chromoside B' with 0.5 N sodium hydroxide gave methyl deacetylchromoside B and isobutyric acid, the latter being identified as its p-bromophenacyl ester. Thus, chromose B' is 4-O-isobutyryldeacetylchromose B (3)<sup>15</sup>, and chromomycin  $A_2$  should be represented by structure 2 (molecular formula  $C_{53}H_{76}O_{23})^{14}$  3] Chromomycin  $A_4$  (3)

Chromomycin A<sub>4</sub> was obtained as a yellow powder with  $[a]_D^{21}$ -47°

(c 1.0 in ethanol). The spectroscopic chromatographic and chemical

properties were indistinguishable from those of CHR-D-C-A (7),

 $C_{42}H_{58}O_{19}^{14}$ , the partial hydrolysis product of  $A_3$ , and therefore the two are identical.

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- 3) References cited in reference 8.
- 4) "Toyomycin", Takeda Chemical Industries.
- 5) Although the chromomycins could not be obtained crystalline, all elementary analyses of chromomycins and derivatives were in good agreement with calculated values. The homogeneity of noncrystalline samples were checked by chromatography.
- 6) M. Miyamoto, K. Morita, Y. Kawamatsu, M. Sasai, A. Nohara, K. Tanaka, S. Tatsuoka, K. Nakanishi, Y. Nakadaira and N. S. Bhacca, <u>Tetrahedron Letters</u>, 2367 (1964).
- 7) CHR stands for the chromophore, chromomycinone<sup>12)</sup>, while A, B, C and D stand for the respective chromoses.
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- 12) To be published .
- 13) Compared with authentic samples; details to be published.
- 14) Elementary analyses of chromomycins  $\rm A_2,\ A_3$  and  $\rm A_4$  showed that all three contained one mole of water.
- 15) Point of attachment of the isobutyryl group was easily disclosed by comparing the NMR spectra of chromose B, deacetylchromose B and chromose B'.

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